

FORANE® 408A

1. PRODUCT AND COMPANY IDENTIFICATION

Company

Arkema Inc. 900 First Avenue King of Prussia, Pennsylvania 19406

Fluorochemicals

Customer Service Telephone Number: (800) 245-5858

(Monday through Friday, 8:00 AM to 5:00 PM EST)

Emergency Information

Transportation: CHEMTREC: (800) 424-9300 (24 hrs., 7 days a week)

Medical: Rocky Mountain Poison Center: (866) 767-5089

(24 hrs., 7 days a week)

Product Information

Product name: FORANE® 408A

Synonyms: R-408A, HFC 408A, FORANE FX 10

Molecular formula:Complex MixtureChemical family:Hydrochlorofluorocarbon

Molecular weight: 87.01 g/mol

Product use: Low temperature refrigerant, Air conditioning

2. HAZARDS IDENTIFICATION

Emergency Overview

Color:Clear - colourlessPhysical state:gaseousForm:Liquefied gasOdor:Slightly ether-like

*Classification of the substance or mixture:

Gases under pressure, Liquefied gas, H280 Hazardous to the ozone layer, Category 1, H420

*For the full text of the H-Statements mentioned in this Section, see Section 16.

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GHS-Labelling

Hazard pictograms:





Signal word: Warning

Hazard statements:

H280: Contains gas under pressure; may explode if heated.

H420: Harms public health and the environment by destroying ozone in the upper atmosphere.

Supplemental Hazard Statements:

Overheating or overpressurizing may cause gas release or violent cylinder bursting. May decompose on contact with flames or extremely hot metal surfaces to produce toxic and corrosive products. May cause frostbite. May cause headache, nausea, dizziness, drowsiness, loss of consciousness. May cause cardiac sensitization/cardiac arrhythmia. May displace oxygen and cause rapid suffocation.

Precautionary statements:

Storage:

P403: Store in a well-ventilated place.

P410: Protect from sunlight.

Disposal:

P502: Refer to manufacturer/ supplier for information on recovery/ recycling.

Supplemental information:

Potential Health Effects:

Liquid: Contact with liquid or refrigerated gas can cause cold burns and frostbite. Vapor: Vapor is heavier than air and can cause suffocation by reducing oxygen available for breathing. If inhaled: Central nervous system effects: headache, nausea, dizziness, drowsiness, loss of consciousness. Stress induced heart effects: Inhalation may cause an increase in the sensitivity of the heart to adrenaline, which could result in irregular or rapid heartbeats and reduced heart function.

Medical conditions aggravated by overexposure:

Heart disease or compromised heart function.

3. COMPOSITION/INFORMATION ON INGREDIENTS

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Chemical Name	CAS-No.	Wt/Wt	GHS Classification**
Methane, chlorodifluoro-	75-45-6	>= 30 - < 60 %	H280, H420
Ethane, 1,1,1-trifluoro-	420-46-2	>= 30 - < 60 %	H220, H280
Ethane, pentafluoro-	354-33-6	>= 5 - < 10 %	H280

^{**}For the full text of the H-Statements mentioned in this Section, see Section 16.

4. FIRST AID MEASURES

Inhalation:

If inhaled, remove victim to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

Skin:

If on skin, flush exposed skin with lukewarm water (not hot), or use other means to warm skin slowly. Get medical attention if frostbitten by liquid or if irritation occurs. Remove contaminated clothing and shoes. Wash clothing before reuse. Thoroughly clean shoes before reuse.

Eyes:

Immediately flush eye(s) with plenty of water.

Ingestion:

Ingestion is not applicable - product is a gas at ambient temperatures.

Notes to physician:

Do not give drugs from adrenaline-ephedrine group.

5. FIREFIGHTING MEASURES

Extinguishing media (suitable):

Use extinguishing media appropriate to surrounding fire conditions.

Protective equipment:

Fire fighters and others who may be exposed to products of combustion should wear full fire fighting turn out gear (full Bunker Gear) and self-contained breathing apparatus (pressure demand / NIOSH approved or equivalent).

Further firefighting advice:



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Fight fire with large amounts of water from a safe distance.

Stop the flow of gas if possible.

Water mist should be used to reduce vapor concentrations in air.

Cool closed containers exposed to fire with water spray.

Closed containers of this material may explode when subjected to heat from surrounding fire.

After a fire, wait until the material has cooled to room temperature before initiating clean-up activities.

Fire fighting equipment should be thoroughly decontaminated after use.

Fire and explosion hazards:

May decompose on contact with flames or extremely hot metal surfaces to produce toxic and corrosive products. Liquid and gas under pressure, overheating or overpressurizing may cause gas release and/or violent cylinder bursting.

Container may explode if heated due to resulting pressure rise.

Some mixtures of HCFCs and/or HFCs, and air or oxygen may be combustible if pressurized and exposed to extreme heat or flame.

When burned, the following hazardous products of combustion can occur:

hydrofluoric acid

Carbon oxides

Carbonyl halides

6. ACCIDENTAL RELEASE MEASURES

In case of spill or leak:

Prevent further leakage or spillage if you can do so without risk. Evacuate area of all unnecessary personnel. Eliminate all ignition sources. Use Halogen leak detector or other suitable means to locate leaks or check atmosphere. Keep upwind. Evacuate enclosed spaces and disperse gas with floor-level forced-air ventilation. Avoid breathing leaked material. Consult a regulatory specialist to determine appropriate state or local reporting requirements, for assistance in waste characterization and/or hazardous waste disposal and other requirements listed in pertinent environmental permits.

7. HANDLING AND STORAGE

Handling

General information on handling:

Avoid breathing gas.

Avoid contact with skin, eyes and clothing.

Keep away from heat, sparks and flames.

Wear cold-insulating gloves/face shield/eye protection.

Keep container closed.

Use only with adequate ventilation.

Use equipment rated for cylinder pressure.

Use a backflow preventative device in piping.

Wash thoroughly after handling.

Close valve after each use and when empty.

Do not enter confined spaces unless adequately ventilated.

DO NOT CUT, DRILL, GRIND, OR WELD ON OR NEAR THIS CONTAINER.

Emptied container retains vapor and product residue.

Observe all labeled safeguards until container is cleaned, reconditioned or destroyed.

Storage

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General information on storage conditions:

Keep away from direct sunlight. Keep cylinders restrained. Store in cool, dry, well ventilated area away from sources of ignition such as flame, sparks and static electricity.

Storage stability - Remarks:

Do not apply direct flame to cylinder. Do not store cylinder in direct sun or expose it to heat above 120 F (48.9 C.). Do not drop or refill this cylinder.

Storage incompatibility - General:

Store separate from:

Finely divided metals (aluminium, magnesium, zinc...)

Strong bases

Alkali metals

Alkaline earth metals

Strong oxidizing agents

Temperature tolerance - Do not store above:

118 °F (48 °C)

8. EXPOSURE CONTROLS/PERSONAL PROTECTION

Airborne Exposure Guidelines:

Methane, chlorodifluoro- (75-45-6)

US. ACGIH Threshold Limit Values

Time weighted average 1,000 ppm

Ethane, 1,1,1-trifluoro- (420-46-2)

US. OARS. WEELs Workplace Environmental Exposure Level Guide

Time weighted average 1,000 ppm (3,400 mg/m3)

Remarks: Listed

Ethane, pentafluoro- (354-33-6)

US. OARS. WEELs Workplace Environmental Exposure Level Guide

Time weighted average 1,000 ppm (4,900 mg/m3)

Remarks: Listed

Only those components with exposure limits are printed in this section. Limits with skin contact designation above have skin contact effect. Air sampling alone is insufficient to accurately quantitate exposure. Measures to prevent significant cutaneous absorption may be required. Limits with a sensitizer designation above mean that exposure to this material may cause allergic reactions.

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Engineering controls:

Investigate engineering techniques to reduce exposures below airborne exposure limits or to otherwise reduce exposures. Provide ventilation if necessary to minimize exposures or to control exposure levels to below airborne exposure limits (if applicable see above). If practical, use local mechanical exhaust ventilation at sources of air contamination such as open process equipment.

Monitor carbon monoxide and oxygen levels in tanks and enclosed spaces. Consult ACGIH ventilation manual, NFPA Standard 91 and NFPA Standard 654 for design of exhaust system and safe handling.

Respiratory protection:

Avoid breathing gas. Where airborne exposure is likely or airborne exposure limits are exceeded (if applicable, see above), use NIOSH approved respiratory protection equipment appropriate to the material and/or its components (full facepiece recommended). Consult respirator manufacturer to determine appropriate type equipment for a given application. Observe respirator use limitations specified by NIOSH or the manufacturer. For emergency and other conditions where there may be a potential for significant exposure or where exposure limit may be significantly exceeded, use an approved full face positive-pressure, self-contained breathing apparatus or positive-pressure airline with auxiliary self-contained air supply. Respiratory protection programs must comply with 29 CFR § 1910.134.

Skin protection:

Wear appropriate chemical resistant protective clothing and chemical resistant gloves to prevent skin contact. Consult glove manufacturer to determine appropriate type glove material for given application. Rinse immediately if skin is contaminated. Wash contaminated clothing and clean protective equipment before reuse. Wash thoroughly after handling.

Eye protection:

Use good industrial practice to avoid eye contact.

9. PHYSICAL AND CHEMICAL PROPERTIES Color: Clear - colourless Physical state: gaseous Form: Liquefied gas Odor: Slightly ether-like Odor threshold: not determined Flash point Not applicable **Auto-ignition** not determined temperature: Lower flammable limit None. (LFL): Upper flammable limit None. (UFL):

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pH: Not applicable

Density: not determined

Specific Gravity (Relative

density):

1.06 (77 °F(25 °C))

Vapor pressure: 7,834 mmHg (70.0 °F (21.1 °C))

Vapor density: 3.02 kg/m3

Boiling point/boiling

range:

-46.3 °F (-43.5 °C)

Freezing point: not determined

Melting point: not determined

Evaporation rate: No data available

Solubility in water: Slightly soluble

% Volatiles: 100 %

Molecular weight: 87.01 g/mol

Oil/water partition

coefficient:

No data available

Thermal decomposition No data available

Flammability: See GHS Classification in Section 2

10. STABILITY AND REACTIVITY

Stability:

This material is chemically stable under normal and anticipated storage, handling and processing conditions.

Materials to avoid:

Alkaline earth metals Strong oxidizing agents Finely divided metals (aluminium, magnesium, zinc...) Alkali metals Strong bases

Conditions / hazards to avoid:

Heat

Hazardous decomposition products:

Thermal decomposition giving toxic and corrosive products:

Hydrogen fluoride Carbonyl halides

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Carbon oxides

11. TOXICOLOGICAL INFORMATION

Data on this material and/or its components are summarized below.

Data for Methane, chlorodifluoro- (75-45-6)

Acute toxicity

Inhalation:

Practically nontoxic. (Rat) 4 h LC50 220000 ppm. (Gas)

Skin Irritation:

Practically non-irritating. (Rabbit) (Rapid evaporation of the liquid may cause frostbite.)

Eye Irritation:

Causes mild eye irritation. (Rabbit) (30 s) signs: Rapid evaporation of the liquid may cause frostbite (gas spray)

Sensitization:

Causes cardiac sensitization. (dog, rat, mouse, rabbit and monkey) irregular heart beat, rapid heart beat, in some cases, sudden death

Skin Sensitization:

Not a sensitizer. Repeated skin exposure. (Guinea pig) No skin allergy was observed

Repeated dose toxicity

Chronic inhalation administration to rat, mouse / No adverse systemic effects reported.

Chronic oral administration to Rat / No adverse systemic effects reported.

Carcinogenicity

Chronic inhalation administration to mice / signs: No increase in tumor incidence was reported.

Chronic inhalation administration to female rat / signs: No increase in tumor incidence was reported.

Chronic inhalation administration to male rat / affected organ(s): salivary gland / signs: Increased incidence of tumors was reported.

Genotoxicity

Assessment in Vitro:

Genetic changes were observed in laboratory tests using: bacteria

No genetic changes were observed in laboratory tests using: animal cells, yeast

Genotoxicity

Assessment in Vivo:

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No genetic changes were observed in laboratory tests using: mice

Developmental toxicity

Exposure during pregnancy. inhalation (Rat) / Birth defects were observed. (eye) Exposure during pregnancy. inhalation (Rabbit) / No birth defects were observed.

Reproductive effects

Reproduction test. inhalation (rat and mouse) / No toxicity to reproduction / (males)

Human experience

Inhalation:

Lung: Asphyxia, suffocation.

Heart: Palpitation. (based on reports of occupational exposure to workers)

Human experience

Skin contact:

Skin: irritation, redness, swelling. (repeated or prolonged exposure)

Data for Ethane, 1,1,1-trifluoro- (420-46-2)

Acute toxicity

Inhalation:

No deaths occurred. (Rat) 4 h LC0 > 591000 ppm.

Sensitization:

Causes cardiac sensitization. Inhalation. (Dog) Stress induced heart effects: irregular heart beat, rapid heart beat, in some cases, sudden death (Reaction may occur in response to stress (natural adrenaline release) or administration of epinephrine.)

Repeated dose toxicity

Repeated inhalation administration to rat and guinea pig / affected organ(s): lung / signs: irritation, bronchitis, pneumonia

Chronic oral administration to rat / No adverse effects reported.

Carcinogenicity

Chronic oral administration to rat / signs: No increase in tumor incidence was reported.

Genotoxicity

Assessment in Vitro:

No genetic changes were observed in laboratory tests using: bacteria, human cells

Genotoxicity

Assessment in Vivo:

No genetic changes were observed in laboratory tests using: mice

Developmental toxicity

Exposure during pregnancy. Inhalation (rat and rabbit) / No birth defects were observed.

Data for Ethane, pentafluoro- (354-33-6)

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Acute toxicity

Inhalation:

Practically nontoxic. (Rat) 4 h LC50 > 800000 ppm. (Gas)

Sensitization:

Causes cardiac sensitization. inhalation. (Dog) Stress induced heart effects: irregular heart beat, rapid heart beat, in some cases, sudden death (Reaction may occur in response to stress (natural adrenaline release) or administration of epinephrine.)

Repeated dose toxicity

Subchronic inhalation administration to Rat / No adverse systemic effects reported.

Genotoxicity

Assessment in Vitro:

No genetic changes were observed in laboratory tests using: bacteria, animal cells, human cells

Genotoxicity

Assessment in Vivo:

No genetic changes were observed in laboratory tests using: mice

Developmental toxicity

Exposure during pregnancy. inhalation (rat and rabbit) / No birth defects were observed.

12. ECOLOGICAL INFORMATION

Chemical Fate and Pathway

Data on this material and/or its components are summarized below.

Data for Methane, chlorodifluoro- (75-45-6)

Biodegradation:

Not readily biodegradable. (28 d) biodegradation 0 %

Octanol Water Partition Coefficient:

log Pow = 1.08 (Practically no potential to bioaccumulate.)

Photodegradation:

Half-life direct photolysis: = 8.4 y

Mobility and Distribution in the Environment:

Moderate adsorption / Log Koc = 1.8

Global Warming Potential:

GWP 1,810 (Global warming potential with respect to CO2 (time horizon 100 years)) GWP 0.33 (Halocarbon global warming potential; HGWP; (R-11 = 1))

Ozone Depletion Potential:

ODP 0.055 (Ozone depletion potential; ODP; (R-11 = 1))

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Data for Ethane, 1,1,1-trifluoro- (420-46-2)

Biodegradation:

Not readily biodegradable. (28 d) biodegradation 3 - 10 % / similar material

Octanol Water Partition Coefficient:

log Pow = 1.73 (calculated)

Global Warming Potential:

GWP 3,800 (Global warming potential with respect to CO2 (time horizon 100 years))

Ozone Depletion Potential:

ODP 0 (Ozone depletion potential; ODP; (R-11 = 1))

Data for Ethane, pentafluoro- (354-33-6)

Biodegradation:

Not readily biodegradable. (Closed Bottle test, 28 d) biodegradation 5 %

Octanol Water Partition Coefficient:

log Pow = 1.48

Global Warming Potential:

GWP 0.84 (Halocarbon global warming potential; HGWP; (R-11 = 1)) GWP 3,450 (Global warming potential with respect to CO2 (time horizon 100 years))

Ozone Depletion Potential:

ODP 0.001 (Ozone depletion potential; ODP; (R-11 = 1))

Ecotoxicology

Data on this material and/or its components are summarized below.

Data for Methane, chlorodifluoro- (75-45-6)

Aquatic toxicity data:

Practically nontoxic. Brachydanio rerio (zebrafish) 96 h LC50 = 777 mg/l

Aquatic invertebrates:

Practically nontoxic. Daphnia magna (Water flea) 48 h EC50 = 433 mg/l

Microorganisms:

Bacteria 24 h Toxicity threshold > 400 mg/l (under anaerobic conditions)

Data for Ethane, 1,1,1-trifluoro- (420-46-2)

Aquatic toxicity data:

No adverse effects reported. Oncorhynchus mykiss (rainbow trout) 96 h LC0 >= 175 mg/l (Nominal concentration)

Aquatic invertebrates:

Practically nontoxic. Daphnia magna (Water flea) 48 h EC50 = 300 mg/l

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13. DISPOSAL CONSIDERATIONS

Waste disposal:

Do not vent the container contents, or product residuals, to the atmosphere. Recover and reclaim unused contents or residuals as appropriate. Recovered/reclaimed product can be returned to an approved certified reclaimer or back to the seller depending on the material. Completely emptied disposable containers can be disposed of as recyclable steel. Returnable cylinders must be returned to seller. Dispose of in accordance with federal, state and local regulations. Consult a regulatory specialist to determine appropriate state or local reporting requirements, for assistance in waste characterization and/or hazardous waste disposal and other requirements listed in pertinent environmental permits. Note: Chemical additions to, processing of, or otherwise altering this material may make this waste management information incomplete, inaccurate, or otherwise inappropriate. Furthermore, state and local waste disposal requirements may be more restrictive or otherwise different from federal laws and regulations.

14. TRANSPORT INFORMATION

US Department of Transportation (DOT)

UN Number : 3163

Proper shipping name : Liquefied gas, n.o.s.

Technical name : (Chlorodifluoromethane, 1.1.1-TRIFLUOROETHANE)

Class : 2.2 Marine pollutant : no

International Maritime Dangerous Goods Code (IMDG)

UN Number : 3163

Proper shipping name : LIQUEFIED GAS, N.O.S.

Technical name : (CHLORODIFLUOROMETHANE, 1.1.1-TRIFLUOROETHANE)

Class : 2.2 Marine pollutant : no

15. REGULATORY INFORMATION

Chemical Inventory Status

EU. EINECS EINECS Conforms to

US. Toxic Substances Control Act TSCA The components of this product are all on

the TSCA Inventory.

Australia. Industrial Chemical (Notification and

Assessment) Act

AICS

Conforms to

Canada. Canadian Environmental Protection Act

(CEPA). Domestic Substances List (DSL)

DSL All components of this product are on the

Canadian DSL.

Japan. Kashin-Hou Law List ENCS (JP) Does not conform

Korea. Existing Chemicals Inventory (KECI) KECI (KR) Conforms to

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Philippines. The Toxic Substances and Hazardous and Nuclear Waste Control Act

China. Inventory of Existing Chemical Substances

IECSC (CN)

Conforms to China. Inventory of Existing Chemical Substances

IECSC (CN)

Conforms to China. Inventory of Existing Chemical Substances

United States - Federal Regulations

SARA Title III - Section 302 Extremely Hazardous Chemicals:

The components in this product are either not SARA Section 302 regulated or regulated but present in negligible concentrations.

SARA Title III - Section 311/312 Hazard Categories:

Acute Health Hazard, Sudden Release of Pressure Hazard

SARA Title III - Section 313 Toxic Chemicals:

Chemical Name		De minimis	Reportable threshold:
		concentration	
Methane, chlorodifluoro-	75-45-6	1.0 %	25000 lbs (Manufacturing
			and processing) 10000 lbs (Otherwise used
			(non-
			manufacturing/processing))
			aa.a.ag, p. 6666g//

Comprehensive Environmental Response, Compensation, and Liability Act (CERCLA) - Reportable Quantity (RQ):

The components in this product are either not CERCLA regulated, regulated but present in negligible concentrations, or regulated with no assigned reportable quantity.

United States - State Regulations

New Jersey Right to Know

<u>Chemical Name</u>	CAS-No.
Methane, chlorodifluoro-	75-45-6
Ethane, 1,1,1-trifluoro-	420-46-2



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75-45-6

New Jersey Right to Know - Special Health Hazard Substance(s)

Chemical NameCAS-No.Ethane, 1,1,1-trifluoro-420-46-2

Pennsylvania Right to Know

Chemical NameCAS-No.Methane, chlorodifluoro-75-45-6

Ethane, pentafluoro- 354-33-6

Ethane, 1,1,1-trifluoro- 420-46-2

Pennsylvania Right to Know - Environmentally Hazardous Substance(s)

<u>Chemical Name</u> <u>CAS-No.</u>

California Prop. 65

Methane, chlorodifluoro-

This product does not contain any chemicals known to the State of California to cause cancer, birth defects, or any other reproductive defects.

16. OTHER INFORMATION

Full text of H-Statements referred to under sections 2 and 3.

H220 Extremely flammable gas.

H280 Contains gas under pressure; may explode if heated.

H420 Harms public health and the environment by destroying ozone in the upper atmosphere.

Latest Revision(s):

 Reference number:
 00000057861

 Date of Revision:
 05/15/2015

 Date Printed:
 05/15/2015

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